

This Page Is Inserted by IFW Operations
and is not a part of the Official Record

BEST AVAILABLE IMAGES

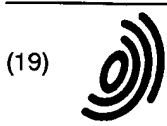
Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images may include (but are not limited to):

- BLACK BORDERS
- TEXT CUT OFF AT TOP, BOTTOM OR SIDES
- FADED TEXT
- ILLEGIBLE TEXT
- SKEWED/SLANTED IMAGES
- COLORED PHOTOS
- BLACK OR VERY BLACK AND WHITE DARK PHOTOS
- GRAY SCALE DOCUMENTS

IMAGES ARE BEST AVAILABLE COPY.

As rescanning documents *will not* correct images,
please do not report the images to the
Image Problem Mailbox.



Europäisches Patentamt
European Patent Office
Office européen des brevets



(11) EP 0 992 485 A1

(12) EUROPEAN PATENT APPLICATION

(43) Date of publication:
12.04.2000 Bulletin 2000/15

(51) Int. Cl.⁷: C07C 69/734, C09K 19/58,
C07D 317/32
// C07M7:00

(21) Application number: 99118675.0

(22) Date of filing: 22.09.1999

(84) Designated Contracting States:
AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU
MC NL PT SE
Designated Extension States:
AL LT LV MK RO SI

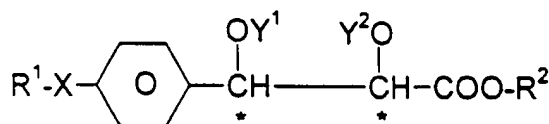
(30) Priority: 02.10.1998 DE 98118646

(71) Applicant: MERCK PATENT GmbH
64293 Darmstadt (DE)

(72) Inventor: Farrand, Louis, Dr.
Davyhulme, Manchester M41 7NT (GB)

(54) Chiral compounds

(57) The invention relates to chiral compounds of formula I



wherein R¹, R², X, Y¹ and Y² have the meaning given in claim 1, to a liquid crystalline mixture comprising at least one chiral compound of formula I, to a chiral linear or crosslinked liquid crystalline polymer obtainable by polymerizing a polymerizable mixture comprising at least one chiral compound of formula I, to the use of chiral compounds of formula I and mixtures and polymers obtained thereof in liquid crystal displays, such as STN, TN, AMD-TN, temperature compensation, guest-host, phase change or surface stabilized or polymer stabilized cholesteric texture (SSCT, PSCT) displays, in active and passive optical elements like polarizers, compensators, alignment layers, colour filters or holographic elements, in adhesives, synthetic resins with anisotropic mechanical properties, cosmetics, diagnostics, liquid crystal pigments, for decorative and security applications, in nonlinear optics, optical information storage or as chiral dopants, and to a liquid crystal display comprising a mixture comprising at least one chiral compound of formula I.

EP 0 992 485 A1

AM

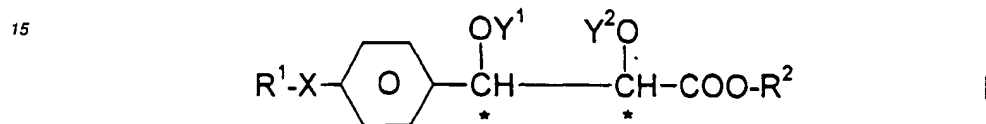
[0015] Another aim of the invention is to extend the pool of chiral compounds that can be used as dopants available to the expert.

[0016] It has been found that these aims can be achieved by providing chiral compounds according to formula I.

[0017] The inventive chiral compounds bear several advantages. Thus, they contain a chiral structure element exhibiting two centres of chirality and thus exhibit a high twisting power. Also, enantiomerically pure compounds of formula I are easy to prepare from cheap, readily available starting materials. The preparation methods are also suitable for large scale production. Furthermore, it is possible to prepare the R,S and S,R enantiomers which can be used to produce a cholesteric phase with either a right or a left handed helix. The availability of both helices can be a considerable advantage e.g. for the use in security film applications.

[0018] The EP 0 441 213-B1 describes chiral compounds comprising a 1,3-dioxolane group attached in 2-position to a mesogenic group, but does not specifically disclose compounds of formula I of the present invention.

[0019] Thus the object of this invention are chiral compounds of formula I



20
wherein

Y¹ is -Y-R³ and Y² is -Y-R⁴, or alternatively Y¹ and Y² form together the bivalent radical -CO- or -C(XR³)(XR⁴)-,

25
R¹, R², R³ and R⁴ are independently of each other H, CN, halogen or an aromatic, aliphatic or araliphatic group with 1 to 50 C atoms,

30
X is in each case independently -O-, -S-, -CH=CH-, -C=C-, -CO-, -COO-, -OCO-, -OCO-O-, -CO-NH-, -NH-CO-, -OCH₂-, -CH₂O-, -SCH₂-, -CH₂S-, -CH=CH-COO-, -OOC-CH=CH- or a single bond and

Y is in each case independently -CO-, -COO-, -CO-NH-, -CO-CH=CH- or a single bond.

35 [0020] Another object of the invention is a liquid crystalline mixture containing at least one chiral compound of formula I.

[0021] Another object of the invention is a polymerizable liquid crystalline mixture comprising at least one chiral compound of formula I and at least one polymerizable mesogenic compound having at least one polymerizable functional group. Another object of the invention is a chiral linear or crosslinked liquid crystalline polymer obtainable by polymerizing such a polymerizable mixture.

40 [0022] A further object of the invention is the use of a chiral compound, mixture or polymer as described above in liquid crystal displays, such as STN, TN, AMD-TN, temperature compensation, guest-host, phase change or surface stabilized or polymer stabilized cholesteric texture (SSCT, PSCT) displays, in active and passive optical elements like polarizers, compensators, alignment layers, colour filters or holographic elements, in adhesives, synthetic resins with anisotropic mechanical properties, cosmetics, diagnostics, liquid crystal pigments, for decorative and security applications, in nonlinear optics, optical information storage or as chiral dopants.

[0023] Yet another object of the invention is a liquid crystal display comprising a liquid crystalline mixture or a polymerizable liquid crystalline mixture comprising at least one chiral compound of formula I.

45 [0024] The inventive chiral compounds can additionally be mesogenic or even liquid crystalline, i.e. they can induce or enhance mesophase behaviour for example in admixture with other compounds, or even exhibit one or more mesophases themselves. It is also possible that the inventive compounds show mesophase behaviour only in mixtures with other compounds, or, in case of polymerizable compounds, when being (co)polymerized. Mesogenic inventive chiral compounds are especially preferred.

50 [0025] The groups R¹, R², R³ and R⁴ can be aromatic or aliphatic groups or araliphatic groups, i.e. combinations thereof.

55 [0026] Suitable aliphatic groups are for example straight chain or branched alkyl groups with 1 to 50, in particular 1 to 25 C atoms, which can optionally be mono-, di- or higher substituted by F, Cl or CN, and wherein one or more non-adjacent CH₂ atoms can be replaced by -O-, -S-, -NH-, -CO-, -CH=CH- or -C=C- groups. Further suitable groups are

cycloaliphatic hydrocarbon groups comprising one, two or three mono- or bicyclic systems, in particular five- or six-membered rings, that may also comprise up to three hetero atoms.

[0027] Suitable aromatic groups are for example phenyl, biphenyl, terphenyl or naphthyl groups, wherein the aromatic rings may also comprise up to three hetero atoms, in particular N atoms, and which can be unsubstituted or mono-, di-, tri- or higher substituted with F, Cl, CN, OH, COOH or alkyl, alkoxy, alkylcarbonyl or alkoxy carbonyl groups with up to 7 C atoms.

[0028] Particular preferably the groups R^1 to R^4 are combinations of the aromatic and aliphatic groups as described above. Further preferred are aromatic and aliphatic groups comprising one or more terminal polymerizable groups, in particular acrylate, methacrylate, vinyl, vinyloxy or epoxy groups. Further preferred are compounds of formula I wherein the groups R^1 , R^2 , R^3 and R^4 comprise one or more chiral C atoms in addition to the chiral structure element shown in formula I. Such compounds exhibit particularly high values of the HTP.

[0029] Especially preferred are chiral compounds of formula I, wherein at least one of R^1 , R^2 , R^3 and R^4 is selected of formula I*



wherein

Sp is in each case independently a spacer group with 1 to 20 C atoms,

X has in each case independently one of the meanings of formula I,

n is 0 or 1,

MG is a mesogenic group, and

R is in each case independently H, CN, halogen or a straight-chain or branched alkyl radical with up to 25 C atoms which may be unsubstituted, mono- or polysubstituted by halogen or CN, it being also possible for one or more non-adjacent CH_2 groups to be replaced, in each case independently from one another, by -O-, -S-, -NH-, -N(CH_3)-, -CO-, -COO-, -OCO-, -OCO-O-, -S-CO-, -CO-S- or -C=C- in such a manner that oxygen atoms are not linked directly to one another, or denoting P-(Sp-X)_n , with Sp, X and n having one of the meanings given above and P being a polymerizable group.

[0030] Further preferred are chiral compounds wherein MG is selected of formula II



wherein

Z is -O-, -S-, -CO-, -COO-, -OCO-, -CO-NH-, -NH-CO-, - CH_2CH_2 -, - OCH_2 -, - CH_2O -, - SCH_2 -, - CH_2S -, - CH=CH -, - CH=CH-COO -, - OCO-CH=CH -, -C=C- or a single bond,

A^1 and A^2 are in each case independently selected from

a) 1,4-phenylene in which, in addition, one or more CH groups may be replaced by N,

b) 1,4-cyclohexylene in which, in addition, one or two non-adjacent CH_2 groups may be replaced by O and/or S,

c) 1,3-dioxolane-4,5-diyl,

d) 1,4-cyclohexenylene, 1,4-bicyclo-(2,2,2)-octylene, piperidine-1,4-diyl, naphthalene-2,6-diyl, decahydronaphthalene-2,6-diyl, or 1,2,3,4-tetrahydronaphthalene-2,6-diyl,

it being possible for all these groups to be unsubstituted, mono- or polysubstituted with halogen, cyano or nitro groups or alkyl, alkoxy, alkylcarbonyl or alkoxy carbonyl groups with 1 to 7 C atoms, wherein one or more H atoms may be substituted by F or Cl, and

m is 0, 1, 2 or 3.

[0031] Further preferred are chiral compounds of formula I, wherein at least one of R^1 , R^2 , R^3 and R^4 is denoting R as defined in formula I*.

5 [0032] Further preferred are chiral compounds of formula I wherein

- R is alkyl or alkoxy with 1 to 12 C atoms,
- at least one of the groups R is denoting $P-(Sp-X)_n-$, with P, Sp, X and n having each independently one of the meanings given in formula I*,

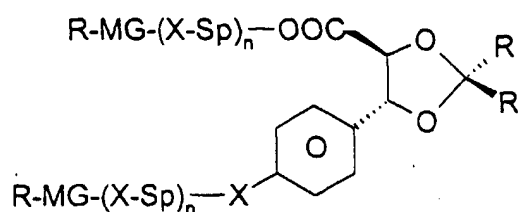
10

[0033] Further preferred are compounds of formula I wherein at least one of, especially both R^1 and R^2 are denoting $-(Sp-X)_n-MG-R$, with P, Sp, X and n having the meaning given in formula I.

[0034] The compounds of formula I are preferably selected of the following formulae

15

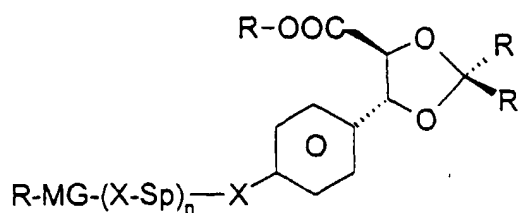
20



Ia

25

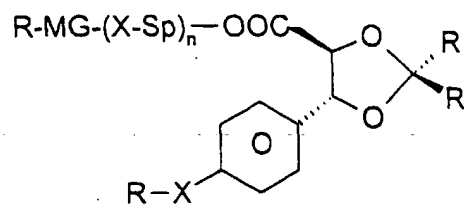
30



Ib

35

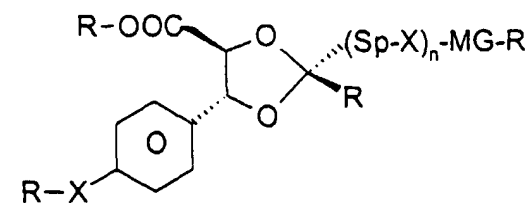
40



Ic

45

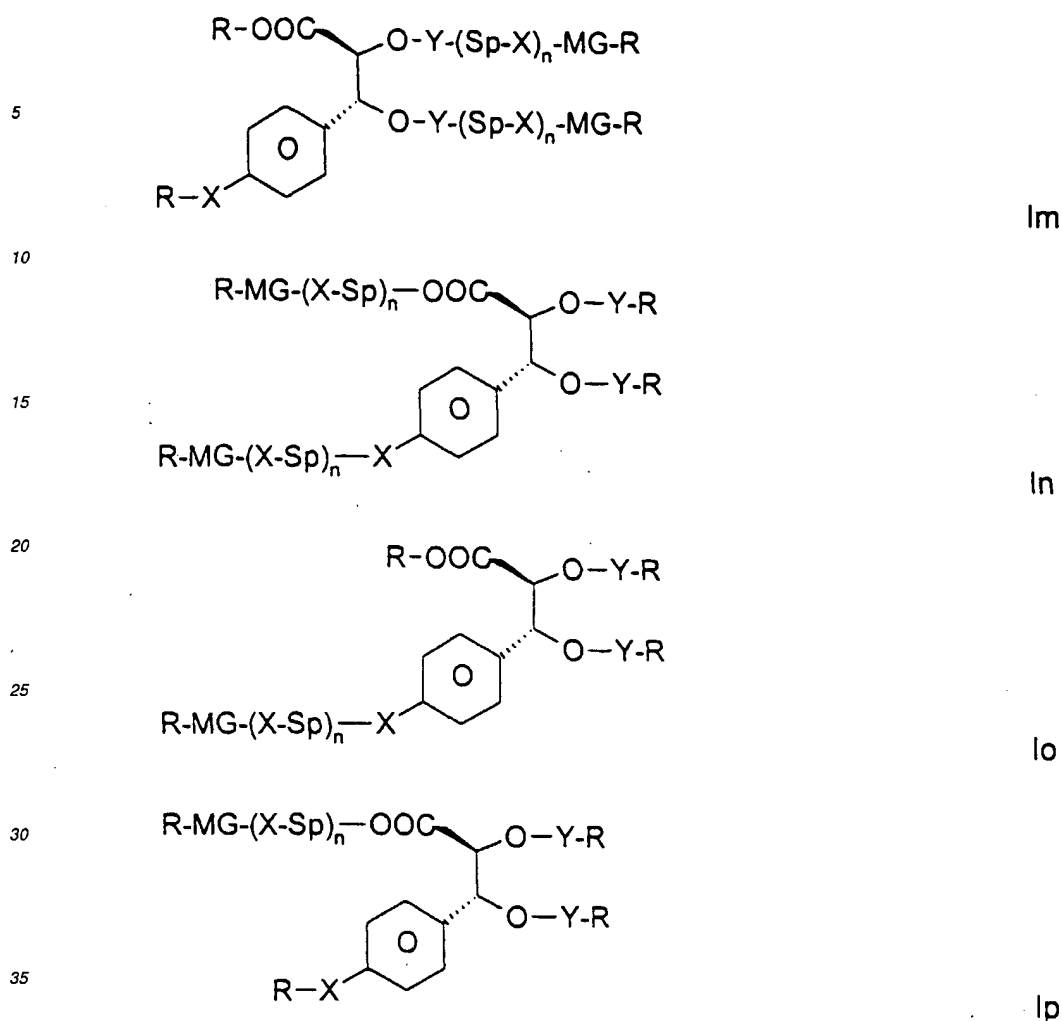
50



Id

55





40 wherein X and Y have each independently one of the meanings of formula I and Sp, MG, n and R have each independently one of the meanings of formula I*.

[0035] Of these preferred compounds particularly preferred are those of formula Ia, Ib, Ic, Id, Im, In and Ip.

[0036] Of the inventive compounds especially preferred are those wherein MG incorporates one, two or three five- or six-membered rings.

45 [0037] Further preferred compounds are those in R is F, Cl, cyano, alkyl or alkoxy with 1 to 12 C atoms and MG is of formula II wherein Z is -COO-, -OCO-, -CH₂-CH₂- or a single bond.

[0038] X is preferably -O-, -CO-, -COO-, -OCO-, -CH₂O-, -OCH₂- or a single bond, in particular -O-, -COO-, -OCO- or a single bond.

[0039] Y is preferably -CO- or a single bond.

50 [0040] Particularly preferred compounds are those wherein MG is of formula II, and A¹ and A² are selected of 1,4-phenylene and trans-1,4-cyclohexylene, these rings being unsubstituted or substituted in 1 to 4 positions with F, Cl, CN or alkyl, alkoxy or alkoxycarbonyl with 1 to 4 C-atoms. From these preferred compounds, especially preferred are those wherein MG is a biphenyl or cyclohexylphenyl group.

[0041] Further preferred compounds are those wherein MG is comprising one or two 1,3-dioxolane-4,5-diyl rings, in particular wherein the 1,3-dioxolane-4,5-diyl groups are linked to an ester group.

55 [0042] A smaller group of preferred mesogenic groups MG of formula II is listed below. For reasons of simplicity, Phe in these groups is 1,4-phenylene, Phe L is a 1,4-phenylene group which is substituted by at least one group L, with L being F, Cl, CN or an optionally fluorinated alkyl, alkoxy or alkanoyl group with 1 to 4 C atoms, and Cyc is 1,4-

EP 0 992 485 A1

cyclohexylene. The list of preferred mesogenic groups is comprising the formulae II-1 to II-27 as well as their mirror images

5	-Phe-	II-1
	-Cyc-	II-2
	-PheL-	II-3
10	-Phe-Z-Phe-	II-4
	-Phe-Z-Cyc-	II-5
15	-Cyc-Z-Cyc-	II-6
	-PheL-Z-Phe-	II-7
	-PheL-Z-Cyc-	II-8
20	-PheL-Z-PheL-	II-9
	-Phe-Z-Phe-Z-Phe-	II-10
	-Phe-Z-Phe-Z-Cyc-	II-11
25	-Phe-Z-Cyc-Z-Phe-	II-12
	-Cyc-Z-Phe-Z-Cyc-	II-13
30	-Phe-Z-Cyc-Z-Cyc-	II-14
	-Cyc-Z-Cyc-Z-Cyc-	II-15
	-Phe-Z-Phe-Z-PheL-	II-16
35	-Phe-Z-PheL-Z-Phe-	II-17
	-PheL-Z-Phe-Z-PheL-	II-18
40	-PheL-Z-PheL-Z-Phe-	II-19
	-PheL-Z-PheL-Z-PheL-	II-20
	-Phe-Z-PheL-Z-Cyc-	II-21
45	-Phe-Z-Cyc-Z-PheL-	II-22
	-Cyc-Z-Phe-Z-PheL-	II-23
50	-PheL-Z-Cyc-Z-PheL-	II-24
	-PheL-Z-PheL-Z-Cyc-	II-25
	-PheL-Z-Cyc-Z-Cyc-	II-26
55	-Cyc-Z-PheL-Z-Cyc-	II-27

[0043] Bicyclic and tricyclic mesogenic groups MG are preferred.

[0044] Further preferred are compounds wherein MG is selected of formula II-7, II-8, II-9 or II-16 to II-27, and L is F, Cl, CH₃, OCH₃, OCF₃ or CN.

[0045] In the above list of preferred groups Z has the meaning given in formula I described above. Preferably Z is -COO-, -OCO-, -CH₂CH₂--CH=CH-COO- or a single bond.

5 **[0046]** L is preferably F, Cl, CN, NO₂, CH₃, C₂H₅, OCH₃, OC₂H₅, COCH₃, COC₂H₅, CF₃, OCF₃, OCHF₂, OC₂F₅, in particular F, Cl, CN, CH₃, C₂H₅, OCH₃, COCH₃ and OCF₃, most preferably F, CH₃, OCH₃ and COCH₃.

[0047] Particularly preferred are chiral compounds wherein MG is selected from the following formulae and their mirror images

10

15

20

25

30

35

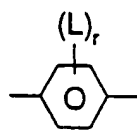
40

45

50

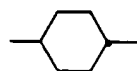
55

5



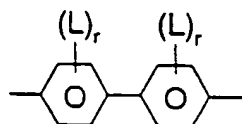
IIa

10



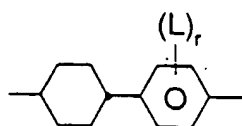
IIb

15



IIc

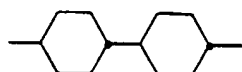
20



25

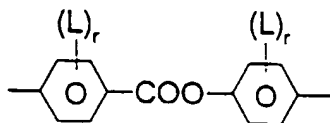
IIId

30



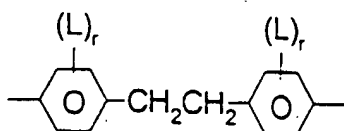
IIe

35



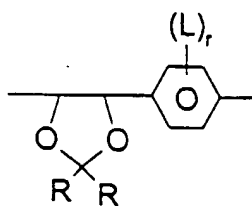
IIIf

40



IIg

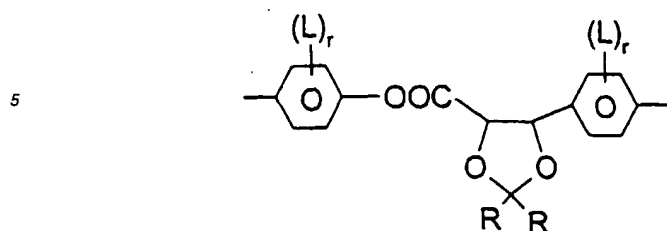
45



50

IIh

55



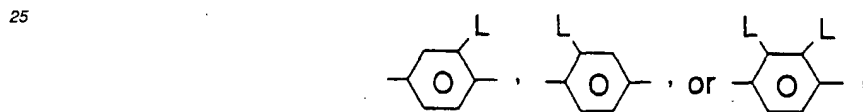
iii

[0048] In these formulae R and L have the meaning given above and r is 0, 1 or 2.

15 [0049] The group



in these preferred formulae is very preferably denoting



30 furthermore



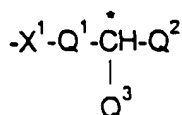
40 with L having each independently one of the meanings given above.

[0050] R in the preferred compounds described above is particularly preferably CN, F, Cl, OCF₃ or an alkyl or alkoxy group with 1 to 12 C atoms. Straight-chain alkyl or alkoxy groups are especially preferred.

[0051] If R is an alkyl or alkoxy radical, i.e. where the terminal CH₂ group is replaced by -O-, this may be straight-chain or branched. It is preferably straight-chain, has 2, 3, 4, 5, 6, 7 or 8 carbon atoms and accordingly is preferably ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, ethoxy, propoxy, butoxy, pentoxy, hexoxy, heptoxy, or octoxy, furthermore methyl, nonyl, decyl, undecyl, dodecyl, tridecyl, tetradecyl, pentadecyl, methoxy, nonoxy, decoxy, undecoxy, dodecoxy, tridecoxy or tetradecoxy, for example.

[0052] Oxaalkyl, i.e. where one CH₂ group is replaced by -O-, is preferably straight-chain 2-oxapropyl (=methoxymethyl), 2- (=ethoxymethyl) or 3-oxabutyl (=2-methoxyethyl), 2-, 3-, or 4-oxapentyl, 2-, 3-, 4-, or 5-oxahexyl, 2-, 3-, 4-, 5-, or 6-oxaheptyl, 2-, 3-, 4-, 5-, 6- or 7-oxaoctyl, 2-, 3-, 4-, 5-, 6-, 7- or 8-oxanonyl or 2-, 3-, 4-, 5-, 6-, 7-, 8- or 9-oxadecyl, for example.

[0053] In the chiral compounds of formula I R may be an achiral or a chiral group. In case of a chiral group it is preferably selected according to the following formula III:



III

10 wherein

X^1 is -O-, -S-, -CO-, -COO-, -OCO-, -OCOO- or a single bond,

Q^1 is an alkylene or alkylene-oxy group with 1 to 10 C atoms or a single bond,

Q^2 is an alkyl or alkoxy group with 1 to 10 C atoms which may be unsubstituted, mono- or polysubstituted by halogen or CN, it being also possible for one or more non-adjacent CH_2 groups to be replaced, in each case independently from one another, by -C=C-, -O-, -S-, -NH-, -N(CH₃)-, -CO-, -COO-, -OCO-, -OCO-O-, -S-CO- or -CO-S- in such a manner that oxygen atoms are not linked directly to one another,

Q^3 is halogen, a cyano group or an alkyl or alkoxy group with 1 to 4 C atoms different from Q^2 .

[0054] Preferred chiral groups R are 2-butyl (=1-methylpropyl), 2-methylbutyl, 2-methylpentyl, 3-methylpentyl, 2-ethylhexyl, 2-propylpentyl, 2-octyl, in particular 2-methylbutyl, 2-methylbutoxy, 2-methylpentoxy, 3-methylpentoxy, 2-ethylhexoxy, 1-methylhexoxy, 2-octyloxy, 2-oxa-3-methylbutyl, 3-oxa-4-methylpentyl, 4-methylhexyl, 2-nonyl, 2-decyl, 2-dodecyl, 6-methoxyoctoxy, 6-methyloctoxy, 6-methyloctanoyloxy, 5-methylheptyloxycarbonyl, 2-methylbutyryloxy, 3-methylvaleroyloxy, 4-methylhexanoyloxy, 2-chloropropionyloxy, 2-chloro-3-methylbutyryloxy, 2-chloro-4-methylvaleryloxy, 2-chloro-3-methylvaleryloxy, 2-methyl-3-oxapentyl, 2-methyl-3-oxahexyl, 1-methoxypropyl-2-oxy, 1-ethoxypropyl-2-oxy, 1-propoxypropyl-2-oxy, 1-butoxypropyl-2-oxy, 2-fluorooctyloxy, 2-fluorodecyloxy for example.

[0055] In addition, chiral compounds of the formula I containing an achiral branched group R may occasionally be of importance, for example, due to a reduction in the tendency towards crystallization. Branched groups of this type generally do not contain more than one chain branch. Preferred achiral branched groups are isopropyl, isobutyl (=methylpropyl), isopentyl (=3-methylbutyl), isopropoxy, 2-methylpropoxy and 3-methylbutoxy.

[0056] A particularly preferred embodiment of the present invention is related to polymerizable compounds of formula I, wherein at least one of the groups R^1 to R^4 is comprising a terminal polymerizable group P.

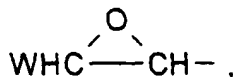
[0057] Of these compounds, especially preferred are those of formulae Ia to Ip, wherein one, two, three or four, in particular one or two groups R are denoting P-(Sp-X)_n . Very particularly preferred are compounds wherein the groups R denoting P-(Sp-X)_n are adjacent to a mesogenic group MG.

[0058] Of the preferred compounds described above in particular preferred are those wherein n is 1.

[0059] Further preferred are compounds comprising at least one group R denoting P-(Sp-X)_n wherein n is 0 and at least one group R denoting P-(Sp-X)_n wherein n is 1.

[0060] Also preferred are compounds wherein at least one, in particular one or two, of R^1 , R^2 , R^3 and R^4 are denoting $\text{-(Sp-X)}_n\text{-MG-R}$ wherein n is 0.

[0061] The polymerizable group P is preferably selected from



$\text{CH}_2=\text{CW-COO-}$, WCH=CH-O- , and $\text{CH}_2=\text{CH-Phenyl-(O)}_k\text{-}$, with W being H, CH_3 or Cl and k being 0 or 1.

[0062] P is preferably a vinyl group, an acrylate group, a methacrylate group, a propenyl ether group or an epoxy group, especially preferably an acrylate or a methacrylate group.

[0063] In the event that one or more groups R^1 to R^4 are denoting $\text{-(Sp-X)}_n\text{-MG-R}$ wherein R is P-(Sp-X)_n , the two spacer groups Sp on both sides of the mesogenic group MG can be identical or different.

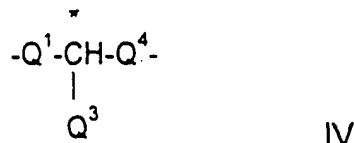
[0064] As for the spacer group Sp in formula I all groups can be used that are known for this purpose to the skilled in the art. The spacer group Sp is preferably a linear or branched alkylene group having 1 to 20 C atoms, in particular

1 to 12 C atoms, in which, in addition, one or more non-adjacent CH₂ groups may be replaced by -O-, -S-, -NH-, -N(CH₃)-, -CO-, -O-CO-, -S-CO-, -O-COO-, -CO-S-, -CO-O-, -CH(halogen)-, -CH(CN)-, -CH=CH- or -C=C-.

[0065] Typical spacer groups are for example -(CH₂)_o-, -(CH₂CH₂O)_r-CH₂CH₂-, -CH₂CH₂-S-CH₂CH₂- or -CH₂CH₂-NH-CH₂CH₂-, with o being an integer from 2 to 12 and r being an integer from 1 to 3.

[0066] Preferred spacer groups are ethylene, propylene, butylene, pentylene, hexylene, heptylene, octylene, nonylene, decylene, undecylene, dodecylene, octadecylene, ethyleneoxyethylene, methyleneoxybutylene, ethylene-thioethylene, ethylene-N-methyliminoethylene, 1-methylalkylene, ethenylene, propenylene and butenylene for example.

[0067] Especially preferred are inventive chiral compounds of formula I wherein Sp is denoting an alkyl or alkoxy group with 2 to 6 C atoms. Straight-chain alkyl or alkoxy groups are especially preferred. In another preferred embodiment of the invention the chiral compounds of formula I comprise at least one spacer group Sp that is a chiral group of the formula IV:



wherein

Q¹ and Q³ have the meanings given in formula III, and

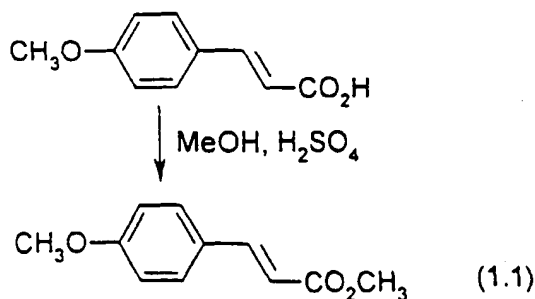
Q⁴ is an alkylene or alkylene-oxy group with 1 to 10 C atoms or a single bond, being different from Q¹.

[0068] The chiral structure element in formula I can be obtained by osmium-catalyzed asymmetric dihydroxylation of the corresponding olefin in the presence of [DHQD]₂PHAL (hydroquinidine 1,4-phthalazinediyl diether) or [DHQ]₂PHAL (hydroquinine 1,4-phthalazinediyl diether, both commercially available from Aldrich), in analogy to the methods described e.g. by K.B. Sharpless et al., J. Org. Chem. 57, 1992, 2768-2771 and S. Torii et al., J. Org. Chem. 61, 1996, 3055-3060. According to this method both the R,S and the S,R enantiomer can be synthesized.

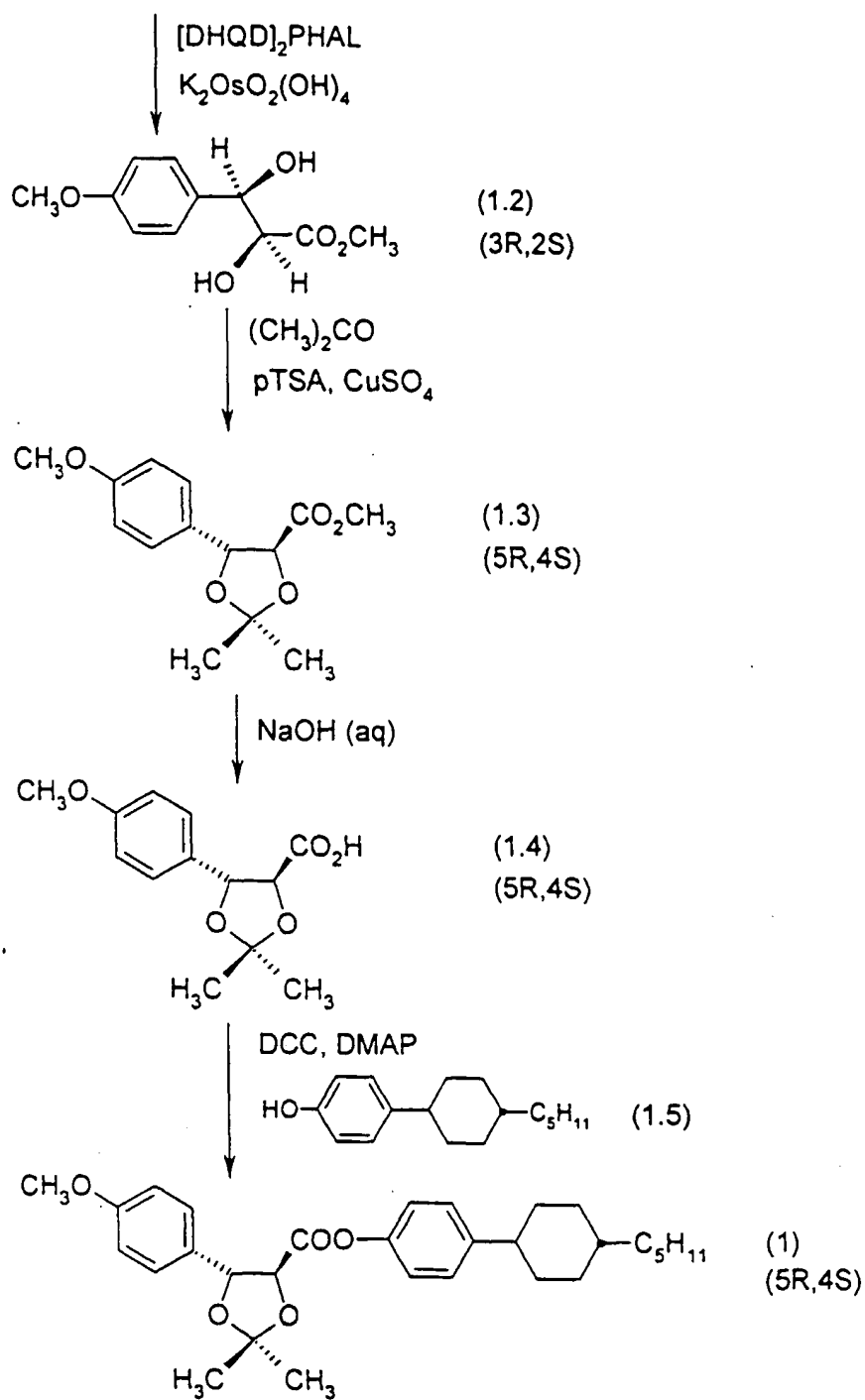
[0069] Especially preferred are chiral compounds of formula I wherein the two chiral C atoms exhibit an S,R- or an R,S-configuration.

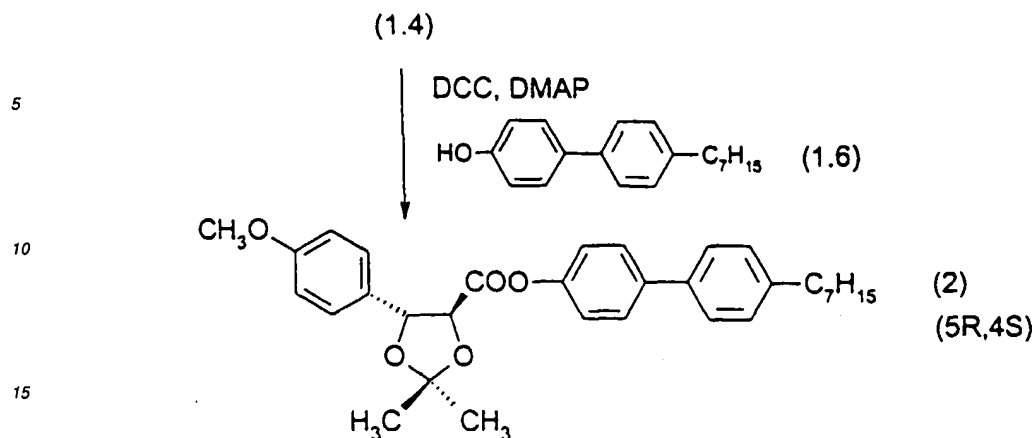
[0070] In particular, the inventive chiral compounds can be synthesized according to or in analogy to reaction scheme 1.

Scheme 1



5
10
15
20
25
30
35
40
45
50
55



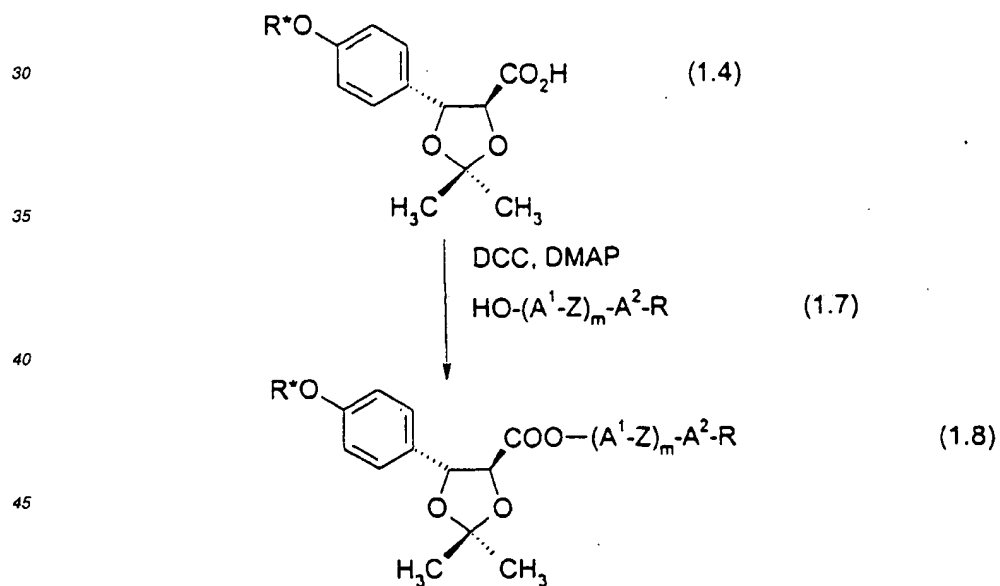


20 wherein DCC is denoting dicyclohexylcarbo-diimide, pTSA is p-toluene sulfonic acid, DMAP is N,N-dimethylamino-pyridine and DCM is dichloromethane.

[0071] Other compounds of formula I can be prepared in analogy to reaction scheme 1. For example, it is possible to react the intermediate (1.4) with a compound of the formula HO-(A¹-Z)_m-A²-R (1.7) according to reaction scheme 2.

25

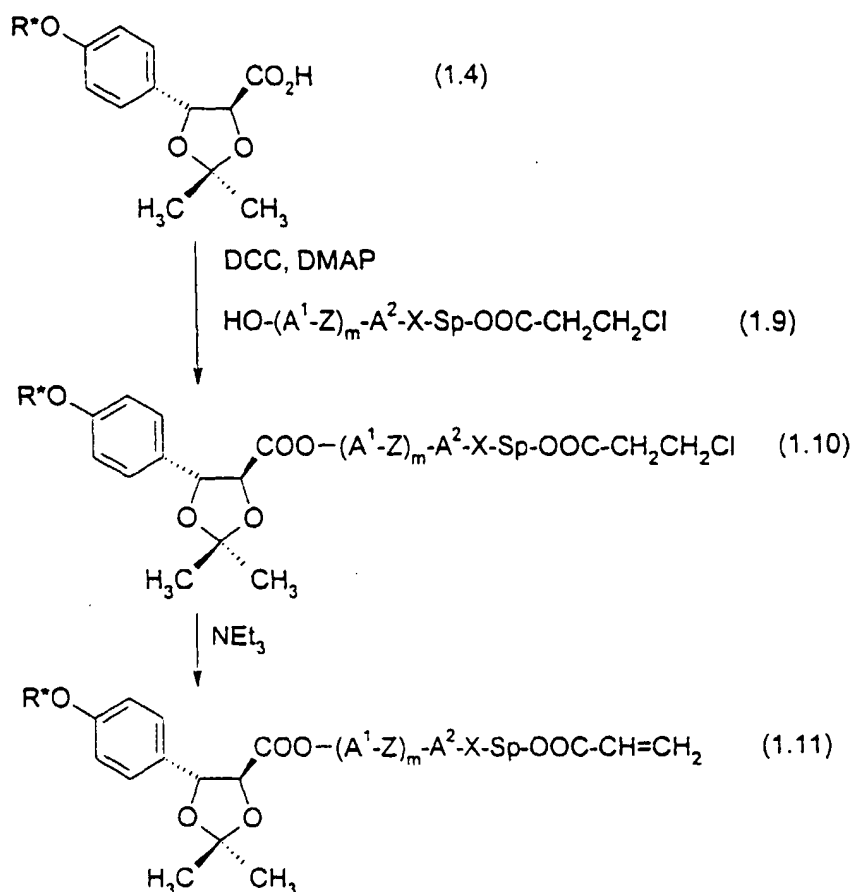
Scheme 2



wherein R^{*} has one of the meanings of R in formula I*, R has the meaning of formula I* and A¹, A², Z and m have the meanings of formula II.

[0072] It is further possible to prepare polymerizable compounds of formula I for example by reacting the intermediate (1.4) with a compound of formula HO-(A¹-Z)_m-A²-X-Sp-OOC-CH₂CH₂Cl (1.9), and treating the resulting intermediate with NEt₃ according to reaction scheme 3.

Scheme 3



wherein R* has the meaning given above, X has the meaning of formula I, Sp has the meaning of formula I* and A¹, A², Z and m have the meanings of formula II.

[0073] Other methods of preparation can be taken from the examples.

[0074] The inventive chiral compounds can be used in a liquid crystal mixture for displays exhibiting a twisted molecular structure of the liquid crystal matrix like, for example, supertwisted or active matrix liquid crystal displays, or in displays comprising a liquid crystal mixture with a chiral liquid crystalline phase, like for example chiral smectic or chiral nematic (cholesteric) mixtures for ferroelectric displays or cholesteric displays.

[0075] Thus, another object of the invention is a liquid crystalline mixture comprising at least one chiral compound of formula I.

[0076] Yet another object of the invention are cholesteric liquid crystal displays comprising cholesteric liquid crystalline media containing at least one chiral compound of formula I.

[0077] The inventive chiral compounds of formula I exhibit high values of the HTP. Thus liquid crystalline mixtures with a high helical twist, i.e. a short cholesteric pitch, can be prepared by using the inventive compounds, or otherwise a liquid crystalline mixture with a moderate helical twist can be achieved already when using the inventive compounds as dopants in low amounts.

[0078] The high HTP values of the inventive compounds makes them also suitable to be used in combination with other compounds for the temperature compensation of the properties of liquid crystal mixtures, such as the cholesteric pitch, and of the properties of displays, e.g. such as the threshold voltage.

[0079] In a preferred embodiment of the invention the chiral compounds show a strong temperature dependence of the HTP in nematic liquid crystal mixtures.

[0080] The inventive compounds are furthermore advantageous because they are affecting the physical properties of the liquid crystalline mixture only to a minor extent.

[0081] Thus, when admixing the chiral compounds of formula I for example to a liquid crystalline mixture with positive dielectric anisotropy that is used in a liquid crystal display, $\Delta\epsilon$ is being only slightly reduced and the viscosity of the liquid crystalline mixture is increased only to a small extent. This leads to lower voltages and improved switching times of the display when compared to a display comprising conventional dopants.

[0082] In a particularly preferred embodiment of the invention the chiral compounds show a small temperature dependence of the HTP in nematic liquid crystal mixtures.

[0083] The liquid crystalline mixture according to the invention comprises preferably 0.001 to 15 %, in particular 0.01 to 8 % and very particularly preferably 0.1 to 5 % by weight of chiral compounds of formula I.

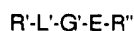
[0084] The liquid crystalline mixture according to the invention preferably comprises 1 to 3 chiral compounds of formula I.

[0085] For temperature compensation applications as described above the liquid crystalline mixture preferably contains a chiral component which contains at least one chiral compound of formula I.

[0086] In a preferred embodiment of the invention the liquid crystalline mixture is consisting of 2 to 25, preferably 3 to 15 compounds, at least one of which is a chiral compound of formula I. The other compounds are preferably low molecular weight liquid crystalline compounds selected from nematic or nematogenic substances, for example from the known classes of the azoxybenzenes, benzylidene-anilines, biphenyls, terphenyls, phenyl or cyclohexyl benzoates, phenyl or cyclohexyl esters of cyclohexanecarboxylic acid, phenyl or cyclohexyl esters of cyclohexylbenzoic acid, phenyl or cyclohexyl esters of cyclohexylcyclohexanecarboxylic acid, cyclohexylphenyl esters of benzoic acid, of cyclohexanecarboxylic acid and of cyclohexylcyclohexanecarboxylic acid, phenylcyclohexanes, cyclohexylbiphenyls, phenylcyclohexylcyclohexanes, cyclohexylcyclohexanes, cyclohexylcyclohexenes, cyclohexylcyclohexylcyclohexenes, 1,4-bis-cyclohexylbenzenes, 4,4'-bis-cyclohexylbiphenyls, phenyl- or cyclohexylpyrimidines, phenyl- or cyclohexylpyridines, phenyl- or cyclohexylpyridazines, phenyl- or cyclohexyldioxanes, phenyl- or cyclohexyl-1,3-dithianes, 1,2-diphenyl-ethanes, 1,2-dicyclohexylethanes, 1-phenyl-2-cyclohexylethanes, 1-cyclohexyl-2-(4-phenylcyclohexyl)-ethanes, 1-cyclohexyl-2-biphenyl-ethanes, 1-phenyl-2-cyclohexyl-phenylethanes, optionally halogenated stilbenes, benzyl phenyl ether, tolanes, substituted cinnamic acids and further classes of nematic or nematogenic substances. The 1,4-phenylene groups in these compounds may also be laterally mono- or difluorinated.

[0087] The liquid crystalline mixture of this preferred embodiment is based on the achiral compounds of this type.

[0088] The most important compounds that are possible as components of these liquid crystalline mixtures can be characterized by the following formula



wherein L' and E, which may be identical or different, are in each case, independently from one another, a bivalent radical from the group formed by -Phe-, -Cyc-, -Phe-Phe-, -Phe-Cyc-, -Cyc-Cyc-, -Pyr-, -Dio-, -B-Phe- and -B-Cyc- and their mirror images, where Phe is unsubstituted or fluorine-substituted 1,4-phenylene, Cyc is trans-1,4-cyclohexylene or 1,4-cyclohexenylene, Pyr is pyrimidine-2,5-diyl or pyridine-2,5-diyl, Dio is 1,3-dioxane-2,5-diyl and B is 2-(trans-1,4-cyclohexyl)ethyl, pyrimidine-2,5-diyl, pyridine-2,5-diyl or 1,3-dioxane-2,5-diyl.

[0089] G' in these compounds is selected from the following bivalent groups -CH=CH-, -N(O)N-, -CH=CY-, -CH=N(O)-, -C=C-, -CH₂-CH₂-, -CO-O-, -CH₂-O-, -CO-S-, -CH₂-S-, -CH=N-, -COO-Phe-COO- or a single bond, with Y being halogen, preferably chlorine, or -CN.

[0090] R' and R'' are, in each case, independently of one another, alkyl, alkenyl, alkoxy, alkenyloxy, alkanoyloxy, alkoxy carbonyl or alkoxy carbonyloxy with 1 to 18, preferably 3 to 12 C atoms, or alternatively one of R' and R'' is F, CF₃, OCF₃, Cl, NCS or CN.

[0091] In most of these compounds R' and R'' are, in each case, independently of each another, alkyl, alkenyl or alkoxy with different chain length, wherein the sum of C atoms in nematic media generally is between 2 and 9, preferably between 2 and 7.

[0092] Many of these compounds or mixtures thereof are commercially available. All of these compounds are either known or can be prepared by methods which are known per se, as described in the literature (for example in the standard works such as Houben-Weyl, Methoden der Organischen Chemie [Methods of Organic Chemistry], Georg-Thieme-Verlag, Stuttgart), to be precise under reaction conditions which are known and suitable for said reactions. Use may also be made here of variants which are known per se, but are not mentioned here.

[0093] The inventive compounds are in particular useful for anisotropic polymer gels and for low molar mass or polymerizable or polymerized cholesteric liquid crystalline mixtures for cholesteric displays, such as for example phase change displays or surface stabilized or polymer stabilized cholesteric texture displays (SSCT, PSCT).

[0094] A further advantage of the chiral compounds according to the invention is that cholesteric liquid crystalline mixtures or materials comprising these compounds exhibit a low temperature dependence of the reflection wavelength

$d\lambda/dT$ (T = temperature, λ = reflection wavelength maximum).

[0095] Cholesteric displays are described for example in WO 92/19695, WO 93/23496, US 5,453,863 or US 5,493,430, with the entire disclosure of these documents being introduced into this application by way of reference.

[0096] Furthermore, anisotropic polymer gels and displays comprising them are disclosed for example in DE 195 04 224 and GB 2 279 659.

[0097] It has been found that PSCT displays comprising the inventive compounds have reduced response times, lower voltages and improved contrast compared to displays comprising conventional dopants, like e.g. R 811 or CB 15, that are commercially available by Merck KGaA (Darmstadt, Germany). For example, PSCT displays in which the conventional dopants are replaced by chiral compounds of according to the present invention can show reduced switching time.

[0098] Cholesteric films made by using the inventive compounds instead of prior art dopants show improved brightness, leading to a better contrast between the coloured planar texture and the almost clear focal conic state which is made black using a black backplate.

[0099] The inventive chiral compounds and polymerizable liquid crystalline mixtures comprising these compounds are also particularly useful for the preparation of anisotropic polymer films with a chiral liquid crystalline phase, such as cholesteric or chiral smectic polymer films, in particular films that exhibit helically twisted molecular structure with uniform planar orientation, i.e. wherein the helical axis is oriented perpendicular to the plane of the film.

[0100] For example, oriented cholesteric polymer films can be used as broad waveband reflective polarizers, as described e.g. in EP 0 606 940, as colour filters, for security markings, or for the preparation of liquid crystal pigments.

I. Heynderickx and D.J. Broer in Mol.Cryst.Liq.Cryst. 203, 113-126 (1991) describe crosslinked cholesteric polymer films that are made of liquid crystalline diacrylates and contain a low molecular weight chiral dopant.

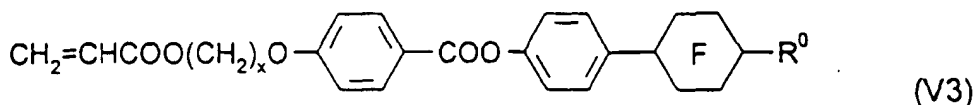
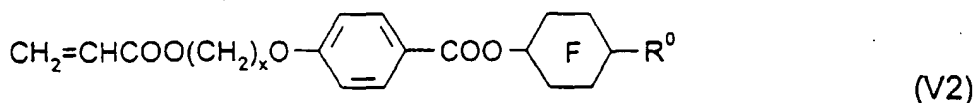
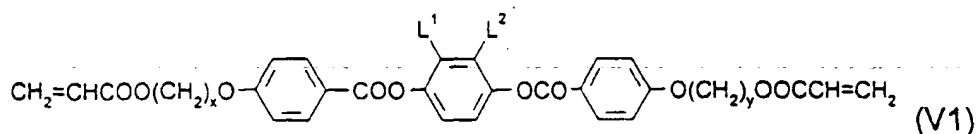
[0101] It has been found that cholesteric polymer films made by using the inventive chiral compounds are brighter compared to films comprising dopants of prior art like e.g. R 811 or CB 15 as mentioned above.

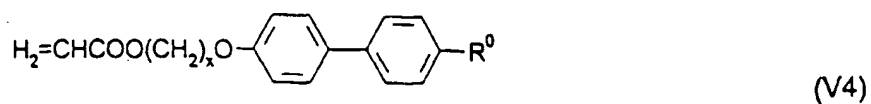
[0102] For the preparation of anisotropic polymer gels or oriented polymer films, the liquid crystalline mixture should comprise at least one polymerizable compound, preferably a polymerizable mesogenic compound, in addition to chiral compounds of formula I.

[0103] Thus, another object of the invention are polymerizable liquid crystalline mixtures comprising at least one chiral compound of formula I and at least one polymerizable mesogenic compound.

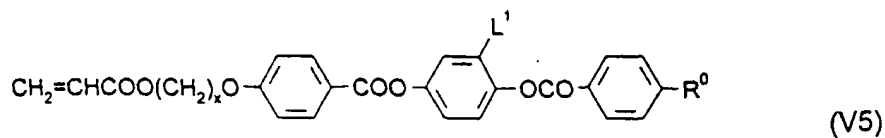
[0104] Suitable polymerizable mesogenic compounds are described for example in WO 93/22397; EP 0 261 712; DE 19504224; DE 4408171 or DE 4405316. The compounds disclosed in these documents, however, are to be regarded merely as examples that shall not limit the scope of this invention.

[0105] Furthermore, typical examples representing polymerizable mesogenic compounds are shown in the following list of compounds, which should, however, be taken only as illustrative and is in no way intended to restrict, but instead to explain the present invention:

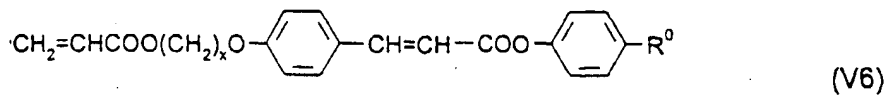




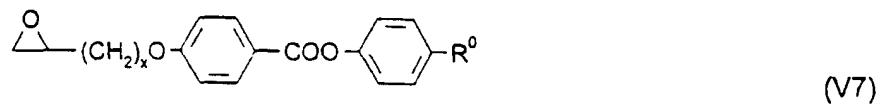
5



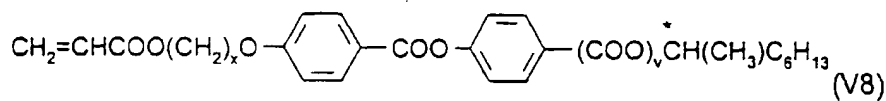
10



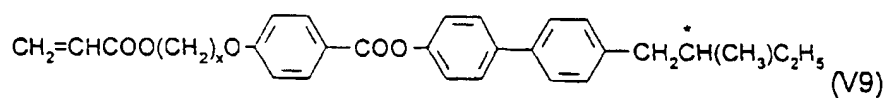
15



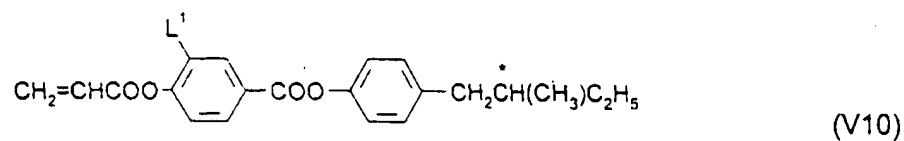
20



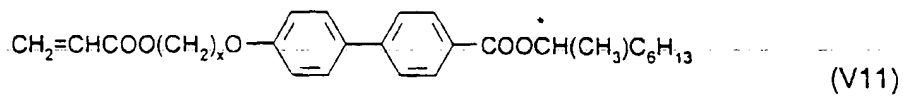
25



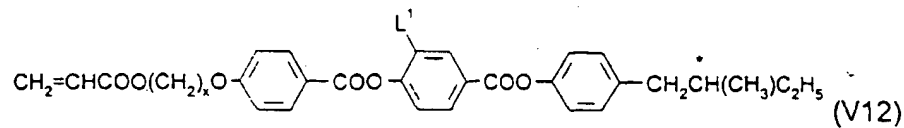
30



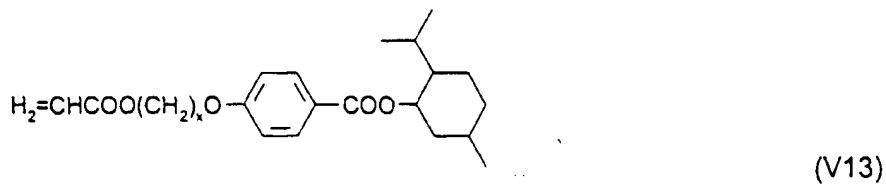
35



40

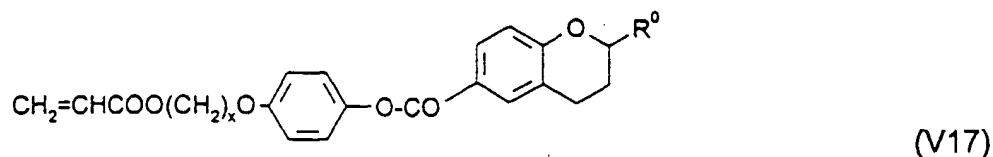
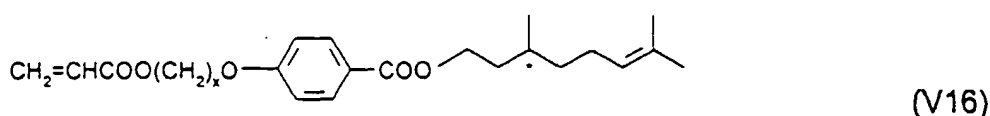
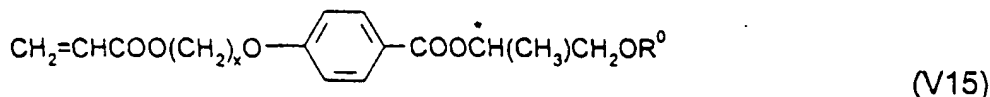
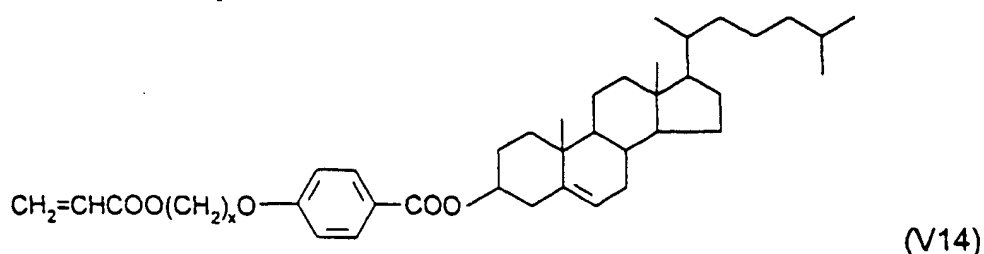


45



50

55



[0106] In these compounds x and y are each independently 1 to 12, F is a 1,4-phenylene or 1,4-cyclohexylene group, R⁰ is halogen, cyano or an optionally halogenated alkyl or alkoxy group with 1 to 12 C atoms and L¹ and L² are each independently H, F, Cl, CN, or an optionally halogenated alkyl, alkoxy or alkoxycarbonyl group with 1 to 7 C atoms.

[0107] The polymerizable mesogenic compounds of formula V1 - V17 can be prepared by methods which are known per se and which are described in the documents cited above and, for example, in standard works of organic chemistry such as, for example, Houben-Weyl, Methoden der organischen Chemie, Thieme-Verlag, Stuttgart.

[0108] The polymerizable mesogenic compounds of formula V1 - V17 can be mono- or bifunctional, i.e. they can have one or two polymerizable functional groups.

[0109] In a preferred embodiment of the invention the polymerizable liquid crystalline mixtures comprise at least one inventive chiral compound, at least one monofunctional and at least one bifunctional polymerizable compound of formulae V1-V17.

[0110] In another preferred embodiment the polymerizable liquid crystalline mixtures comprise at least one inventive chiral compound and at least two monofunctional compounds of formulae V1-V17.

[0111] It is also possible that the polymerizable liquid crystalline mixture comprises one or more polymerizable compounds of formula I instead of or in addition to the polymerizable mesogenic compounds of formulae V1-V17.

[0112] Thus, another object of the invention are polymerizable liquid crystalline mixtures comprising at least one chiral compound of formula I comprising at least one polymerizable functional group.

[0113] In a preferred embodiment the polymerizable liquid crystalline mixtures comprise at least one chiral compound of formula I comprising one polymerizable functional group.

[0114] In another preferred embodiment the polymerizable liquid crystalline mixtures comprise at least one chiral compound of formula I comprising two polymerizable functional groups.

[0115] Another object of the invention is an anisotropic polymer film with an oriented chiral liquid crystalline phase obtainable by (co)polymerizing a liquid crystalline mixture comprising at least one chiral compound of formula I and at least one polymerizable mesogenic compound preferably selected of formula V1-V17 and/or at least one polymerizable chiral compound of formula I.

[0116] To prepare anisotropic polymer film with a chiral liquid crystalline phase with uniform orientation the inventive liquid crystalline mixtures, for example, are coated onto a substrate, aligned and polymerized in situ by exposing them

to heat or actinic radiation. Alignment and curing are preferably carried out in the liquid crystalline phase of the liquid crystalline mixtures.

[0117] Actinic radiation means irradiation with light, like UV light, IR light or visible light, irradiation with X-rays or gamma rays or irradiation with high energy particles, such as ions or electrons. As a source for actinic radiation for example a single UV lamp or a set of UV lamps can be used. Another possible source for actinic radiation is a laser, like e.g. a UV laser, an IR laser or a visible laser.

[0118] For example, when polymerizing by means of UV light, a photoinitiator can be used that decomposes under UV irradiation to produce free radicals or ions that start the polymerization reaction.

[0119] It is also possible to use a cationic photoinitiator, when curing reactive mesogens with for example vinyl and epoxide reactive groups, that photocures with cations instead of free radicals.

[0120] As a photoinitiator for radical polymerization for example the commercially available Irgacure 651, Irgacure 184, Darocure 1173 or Darocure 4205 (all from Ciba Geigy AG) can be used, whereas in case of cationic photopolymerization the commercially available UVI 6974 (Union Carbide) can be used.

[0121] Preferably the polymerizable liquid crystalline mixtures comprising polymerizable chiral compounds of formula I and/or polymerizable mesogenic compounds of formulae V1-V17 additionally comprise 0.01 to 10 %, in particular 0.05 to 8 %, very preferably 0.1 to 5% by weight of a photoinitiator, especially preferably a UV-photoinitiator.

[0122] In a preferred embodiment of the invention the polymerization of the polymerizable mesogenic material is carried out under an atmosphere of inert gas, preferably under a nitrogen atmosphere.

[0123] As a substrate for example a glass or quartz sheet as well as a plastic film or sheet can be used. It is also possible to put a second substrate on top of the coated mixture prior to, during and/or after polymerization. The substrates can be removed after polymerization or not. When using two substrates in case of curing by actinic radiation, at least one substrate has to be transmissive for the actinic radiation used for the polymerization.

[0124] Isotropic or birefringent substrates can be used. In case the substrate is not removed from the polymerized film after polymerization, preferably isotropic substrates are used.

[0125] Preferably at least one substrate is a plastic substrate such as for example a film of polyester such as polyethyleneterephthalate (PET), of polyvinylalcohol (PVA), polycarbonate (PC) or triacetylcellulose (TAC), especially preferably a PET film or a TAC film. As a birefringent substrate for example an uniaxially stretched plastic film can be used. For example PET films are commercially available from ICI Corp. under the trade name Melinex.

[0126] In a preferred embodiment of the present invention, the inventive mixture of the polymerizable liquid crystalline mixture comprising a chiral compound of formula I is coated as a thin layer on a substrate or between substrate, and is preferably aligned in its chiral mesophase, eg. the cholesteric or chiral smectic phase, to give a planar orientation, i.e. an orientation so that the axis of the molecular helix extends transversely to the layer.

[0127] A planar orientation can be achieved for example by shearing the mixture, e.g. by means of a doctor blade. It is also possible to apply an alignment layer, for example a layer of rubbed polyimide or sputtered SiO_x , on top of at least one of the substrates.

[0128] In another preferred embodiment, a second substrate is put on top of the coated material. In this case, the shearing caused by putting together the two substrates is sufficient to give good alignment.

[0129] It is also possible to apply an electric or magnetic field to the coated mixture.

[0130] In some cases it is of advantage to apply a second substrate not only to aid alignment of the polymerizable mixture but also to exclude oxygen that may inhibit the polymerization. Alternatively the curing can be carried out under an atmosphere of inert gas. However, curing in air is also possible using suitable photoinitiators and high lamp power. When using a cationic photoinitiator oxygen exclusion most often is not needed, but water should be excluded.

[0131] A detailed description of the in situ polymerization of polymerizable mesogenic compounds can be found in D.J.Broer et al., Makromolekulare Chemie 190, 2255 (1989).

[0132] The inventive polymerizable liquid crystalline mixtures comprise preferably 0.001 to 15 %, in particular 0.01 to 8 % and very particularly preferably 0.1 to 5 % by weight of non-polymerizable chiral compounds of formula I.

[0133] Polymerizable liquid crystalline mixtures are preferred that comprise 1 to 3 chiral compounds of formula I.

[0134] If polymerizable chiral compounds of formula I are present in the inventive polymerizable liquid crystalline mixtures, these compounds can also be used in the inventive polymerizable liquid crystalline mixtures in higher amounts than given above for the non-polymerizable chiral compounds of formula I.

[0135] In a preferred embodiment of the present invention the polymerizable liquid crystalline mixtures are comprising 1 to 80 % by weight, preferably 2 to 60 %, in particular 5 to 40 % by weight of a polymerizable chiral compound of formula I comprising at least one polymerizable functional group.

[0136] Of the polymerizable liquid crystalline mixtures comprising one or polymerizable chiral compounds of formula I with one polymerizable functional groups (= monofunctional compounds), particularly preferred are those comprising 1 to 60 %, in particular 2 to 45 %, very preferably 3 to 35 % by weight of a polymerizable chiral monofunctional compound of formula I.

[0137] Of the polymerizable liquid crystalline mixtures comprising one or polymerizable chiral compounds of for-

mula I with two polymerizable functional groups (= bifunctional compounds), particularly preferred are those comprising 1 to 50 %, in particular 2 to 35 %, very preferably 3 to 25 % by weight of a polymerizable chiral bifunctional compound of formula I.

[0138] The inventive polymerizable liquid crystalline mixtures can additionally comprise one or more other suitable components, such as, for example, catalysts, sensitizers, stabilizers, co-reacting monomers or surface-active compounds.

[0139] In a preferred embodiment of the invention, the inventive polymerizable liquid crystalline mixture comprises a stabilizer that is used to prevent undesired spontaneous polymerization for example during storage of the composition. As stabilizers in principal all compounds can be used that are known to the skilled in the art for this purpose. These compounds are commercially available in a broad variety. Typical examples for stabilizers are 4-ethoxyphenol or butylated hydroxytoluene (BHT).

[0140] It is also possible, in order to increase crosslinking of the polymers, to add up to 20% of a non mesogenic compound with two or more polymerizable functional groups to the polymerizable composition alternatively or additionally to the multifunctional polymerizable mesogenic compounds.

[0141] Typical examples for difunctional non mesogenic monomers are alkyl diacrylates or alkyl dimethacrylates with alkyl groups of 1 to 20 C atoms. Typical examples for non mesogenic monomers with more than two polymerizable groups are trimethylpropanetrimethacrylate or pentaerythritoltetraacrylate.

[0142] Polymerization of inventive compositions comprising compounds with only one polymerizable functional group leads to linear polymers, whereas in the presence of compounds with more than one polymerizable functional group crosslinked polymers are obtained.

[0143] For the preparation of anisotropic polymer gels, the liquid crystalline mixtures can be polymerized in situ as described above, however, in this case alignment of the polymerizable mixture is not necessary.

[0144] The inventive chiral compounds of formula I can also be used for the preparation of thermochromic liquid crystalline mixtures. Such mixtures are characterized in that they exhibit a chiral liquid crystalline phase or chiral mesophase, like e.g. a chiral smectic phase or a chiral nematic (= cholesteric) phase, with a helically twisted molecular structure that shows selective reflection of a specific waveband of light, wherein the pitch of the molecular helix and thereby the reflected wavelengths are depending on the temperature.

[0145] Especially preferred are inventive liquid crystalline mixtures with thermochromic behaviour that exhibit a cholesteric phase. Of these preferred compositions, further preferred are compositions that exhibit a cholesteric phase and a smectic phase, most preferably a chiral smectic phase, at temperatures below the temperature range of the cholesteric phase. The inventive liquid crystalline mixtures exhibiting thermochromic behaviour can be polymerizable or non-polymerizable.

[0146] From the foregoing description, one skilled in the art can easily ascertain the essential characteristics of this invention, and without departing from the spirit and scope thereof, can make various changes and modifications of the invention to adapt it to various usages and conditions.

[0147] Without further elaboration, it is believed that one skilled in the art can, using the preceding description, utilize the present invention to its fullest extent. The following examples are, therefore, to be construed as merely illustrative and not limitative of the remainder of the disclosure in any way whatsoever.

[0148] In the foregoing and in the following examples, unless otherwise indicated, all temperatures are set forth uncorrected in degrees Celsius and all parts and percentages are by weight.

[0149] The values of the helical twisting power HTP of a chiral compound in a liquid crystalline host are given according to the equation $HTP = (p \cdot c)^{-1}$ in μm^{-1} , wherein p is the pitch of the molecular helix, given in μm , and c is the concentration by weight of the chiral compound in the host given in relative values (thus, e.g. a concentration of 1 % by weight is corresponding to a value of c of 0.01).

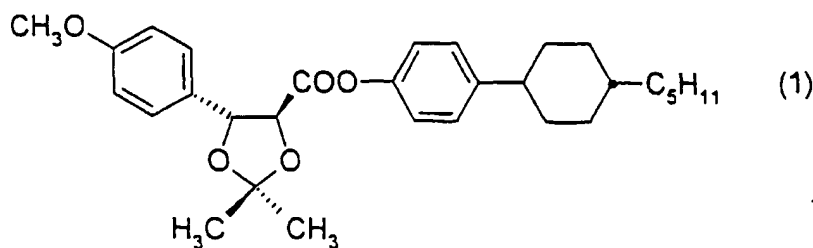
[0150] The following abbreviations are used to illustrate the liquid crystalline phase behaviour of the compounds: K = crystalline; N = nematic; S = smectic; Ch = cholesteric; I = isotropic. The numbers between these symbols indicate the phase transition temperatures in degree Celsius.

[0151] In addition, the following abbreviations are used

DCC = dicyclohexylcarbodiimide
DCM = dichloromethane
HTP = helical twisting power
mp = melting point

Example 1

[0152] Compound (1) was prepared according to reaction scheme 1



(E)-3-(4-methoxyphenyl)-2-propenoate (1.1)

[0153] Trans-Methoxycinnamic acid (17.8 g, 100 mmol), sulphuric acid (2 ml) and methanol (200 ml) were stirred under reflux for 2 hours. The mixture was cooled, concentrated in vacuo, white crystals precipitated and these were isolated by vacuum filtration to give methyl (E)-3-(4-methoxyphenyl)-2-propenoate (16.1 g, 84 %). ¹H-NMR gave the expected signals.

3R,2S-Methyl 2,3-dihydroxy-3-(4-methoxyphenyl)propanoate (1.2)

[0154] [DHQD]₂PHAL (54.5 mg, 0.25 mol %), methyl (E)-3-(4-methoxyphenyl)-2-propenoate (5.38 g, 28 mmol), N-methyl morpholine N-oxide (60% aqueous solution, 7.0 ml) and tert-butanol (11.2 ml) were charged to a 100 ml 3-neck flask and stirred at room temperature. Potassium osmate (VI) dihydrate (21.4 mg, 0.2 mol %) was added portionwise and the reaction was stirred at room temperature. The reaction procedure was monitored by HPLC. After 17 hours, Tiron (200 mg) was added to quench the reaction, followed by water (50 ml). The mixture was allowed to stir for one hour, ethyl acetate (50 ml) was added to the mixture and the resultant two layers were vigorously stirred for 30 minutes. The organic phase was removed and washed with water, dried (Na₂SO₄) and evaporated to dryness on a rotary evaporator. The product was purified by flash column chromatography using DCM: ethyl acetate (3:2) as eluant. Evaporation of the appropriate fractions gave methyl 2,3-dihydroxy-3-(4-methoxyphenyl)propanoate as a white crystalline solid (3.2 g, 62%). mp. 108°C, [α]_D = -1.2° (THF, conc. = 0.01306, 22°C). ¹H-NMR was in accordance with expected signals.

Methyl 5-(4-methoxyphenyl)-2,2-dimethyl-1,3-dioxolane 4-carboxylate (1.3)

[0155] Methyl 2,3-dihydroxy-3-(4-methoxyphenyl)propanoate (3.5 g, 15.5 mmol), copper sulphate anhydrous (4.3 g, 27 mmol, 1.8 equiv.) and acetone (40 ml, excess) were stirred in the presence of a catalytic amount of para-toluene sulphonic acid at 35°C. After 16 hours, the copper sulphate was removed by filtration, and the filtrate was removed and evaporated to dryness on a rotary evaporator. The product was purified by flash column chromatography using DCM as eluant. Evaporation of the appropriate fractions left methyl 5-(4-methoxyphenyl)-2,2-dimethyl-1,3-dioxolane 4-carboxylate as a clear, colourless oil (4.0 g, 96 %). [α]_D = +37.4° (THF, conc. = 0.01712, 20°C). ¹H-NMR was in accordance with expected signals.

(5R,4S)-5-(4-methoxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxylic acid (1.4)

[0156] Methyl 5-(4-methoxyphenyl)-2,2-dimethyl-1,3-dioxolane 4-carboxylate (2.0 g, 7.5 mmol) and aqueous sodium hydroxide solution (0.36 g in 20 ml) and industrial methylated spirits (20 ml) were stirred at 80°C for 3 hours. The mixture was cooled to room temperature, neutralised with dilute hydrochloric acid and concentrated in vacuo. The product was extracted with DCM, dried over Na₂SO₄ and evaporated to dryness to leave (5R,4S)-5-(4-methoxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxylic acid as a colourless oil (1.7 g, 90 %). ¹H-NMR was in accordance with expected signals. I.R. showed expected peaks.

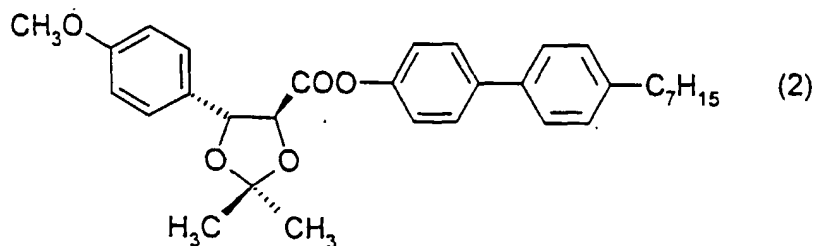
(5R,4S)-5-(4-methoxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxylic acid (1.5)

[0157] (5R,4S)-5-(4-Methoxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxylic acid (1.6 g, 6.3 mmol), DCC (1.44 g, 7.0 mmol), 4-(4'-pentylcyclohexyl)phenol (1.5) (1.55g, 6.3 mmol) and a catalytic amount of dimethylamino pyridine were

stirred at room temperature in dichloromethane (40 ml). After 3 hours, the precipitate of dicyclohexyl urea was removed by filtration, the filtrate was washed with water, dried (Na_2SO_4) and evaporated to dryness. The product was purified by flash column chromatography using DCM as eluant to leave (5R,4S)-5-(4-methoxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxyoxycyclohexyl-4-phenyl-4'-pentane (1), a clear, colourless oil on evaporation of the appropriate fractions. Yield = 1.6 g, 53 %. mp. 56°C . $[\alpha]_D = +89.9^\circ$ (THF, conc. = 0.01124, 25°C). $^1\text{H-NMR}$ was in accordance with expected signals.

Example 2

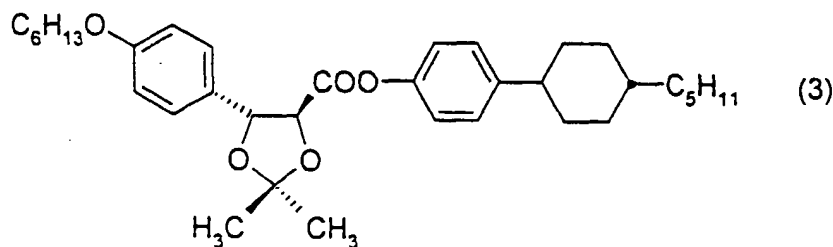
[0158] Compound (2) was prepared according to reaction scheme 1.



[0159] (5R,4S)-5-(4-Methoxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxylic acid (1.4) (0.8 g, 3.2 mmol), DCC (0.7 g, 3.5 mmol), 4-(4'-heptylphenyl)phenol (1.6) (0.8 g, 3.2 mmol) and a catalytic amount of dimethylamino pyridine were stirred at room temperature in dichloromethane (40 ml). After 3 hours, the precipitate of dicyclohexyl urea was removed by filtration, the filtrate was washed with water, dried (Na_2SO_4) and evaporated to dryness. The product was purified by flash column chromatography using DCM as eluant to leave (5R,4S)-5-(4-methoxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxyoxycyclohexyl-4-phenylheptane (2) as a clear, colourless oil on evaporation of the appropriate fractions. Yield = 0.8 g, 50 %. Mp. 60°C . $[\alpha]_D = +103.7^\circ$ (THF, conc. = 0.01326, 25°C). $^1\text{H NMR}$ was in accordance with expected signals.

Example 3

[0160] Compound (3) was prepared according to reaction scheme 1.



[0161] The trans methyl 4-alkoxycinnamate (homologue of compound 1.1) in the first step was prepared by Williamson etherification of trans methyl 4-hydroxycinnamate with the corresponding bromoalkane.

Methyl (E)-3-(4-hexyloxyphenyl)-2-propenoate (3.1)

[0162] Trans methyl 4-hydroxycinnamate (20.0 g, 11.2 mmol), 1-bromohexane (18.5 g, 112 mmol) and potassium carbonate (16.6 g, 120 mmol) were stirred under reflux in butanone. The mixture was allowed to cool after 16 hours, evaporated to dryness and the residue was partitioned between DCM and water. The DCM layer was removed and dried (Na_2SO_4) and evaporated to leave a yellow solid methyl (E)-3-(4-hexyloxyphenyl)-2-propenoate which was recrystallized from petroleum. Yield = 20.0 g, 71 %.

Methyl 2,3-dihydroxy-3-(4-hexyloxyphenyl)propanoate (3.2)

[0163] Synthesis and purification were carried out as described in example 1. Methyl 2,3-dihydroxy-3-(4-hexyloxyphenyl)propanoate was obtained as a colourless oil which crystallized on standing (3.4 g, 62%). ¹H-NMR was in accordance with expected signals.

(5R,4S)-methyl 5-(4-hexyloxyphenyl)-2,2-dimethyl-1,3-dioxolane 4-carboxylate (3.3)

[0164] Synthesis and purification were carried out as described in example 1. (5R,4S)-methyl 5-(4-methoxyphenyl)-2,2-dimethyl-1,3-dioxolane 4-carboxylate was obtained as an oil (2.5 g, 69 %). $[\alpha]_D = +36.5^\circ$ (THF, conc. = 0.01768, 19°C). ¹H-NMR was in accordance with expected signals. The HTP was $2 \mu\text{m}^{-1}$, measured in the commercially available nematic host mixture BL 087 (from Merck Ltd., Poole, UK) at a concentration of 6.34 % by weight.

(5R,4S)-5-(4-hexyloxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxylic acid (3.4)

[0165] Synthesis and purification were carried out as described in example 1. (5R,4S)-5-(4-hexyloxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxylic acid was obtained as a yellow oil (1.7 g, 90%). ¹H-NMR and I.R. spectra were in accordance with expected signals. $[\alpha]_D = +22.5^\circ$ (THF, conc. = 0.01731, 19°C).

(5R,4S)-5-(4-hexyloxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxyloxycyclohexyl-4-phenyl-4'-pentane (3)

[0166] Synthesis and purification were carried out as described in example 1. (5R,4S)-5-(4-methoxyphenyl)2,2-dimethyl-1,3-dioxolane-4-carboxyloxycyclohexyl-4-phenyl-4'-pentane (3) was obtained as a clear oil. Yield = 59 %. $[\alpha]_D = +112.8^\circ$ (THF, conc. = 0.01037, 22°C). ¹H-NMR was in accordance with expected signals.

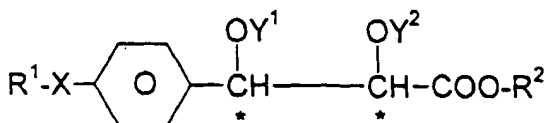
[0167] The HTP of (3) was $14.9 \mu\text{m}^{-1}$, measured in the commercially available nematic host mixture BL 087 (from Merck Ltd., Poole, UK) at a concentration of 5.11 % by weight.

[0168] The preceding examples can be repeated with similar success by substituting the generically or specifically described reactants and/or operating conditions of this invention for those used in the preceding examples.

[0169] From the foregoing description, one skilled in the art can easily ascertain the essential characteristics of this invention, and without departing from the spirit and scope thereof, can make various changes and modifications of the invention to adapt it to various conditions and usages.

Claims

1. Chiral compounds of formula I



wherein

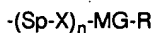
Y^1 is $-Y-R^3$ and Y^2 is $-Y-R^4$, or alternatively Y^1 and Y^2 form together the bivalent radical $-CO-$ or $-C(XR^3)(XR^4)-$.

R^1 , R^2 , R^3 and R^4 are independently of each other H, CN, halogen or an aromatic, aliphatic or araliphatic group with 1 to 50 C atoms,

X is in each case independently $-O-$, $-S-$, $-CO-$, $-COO-$, $-OCO-$, $-OCO-O-$, $CO-NH-$, $-NH-CO-$, $-OCH_2-$, $-CH_2O-$, $-CH=CH-$, $-C=C-$, $-SCH_2-$, $-CH_2S-$, $-CH=CH-COO-$, $-OOC-CH=CH-$ or a single bond, and

Y is in each case independently $-CO-$, $-COO-$, $-CO-NH-$, $-CO-CH=CH-$ or a single bond.

2. Chiral compounds according to claim 1, wherein at least one of R^1 , R^2 , R^3 and R^4 is selected of formula I'



I*

wherein

Sp is in each case independently a spacer group with 1 to 20 C atoms,

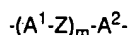
X has in each case independently one of the meanings of formula I,

n is 0 or 1,

MG is a mesogenic group

R is in each case independently H, CN, halogen or a straight-chain or branched alkyl radical with up to 25 C atoms which may be unsubstituted, mono- or polysubstituted by halogen or CN, it being also possible for one or more non-adjacent CH₂ groups to be replaced, in each case independently from one another, by -O-, -S-, -NH-, -N(CH₃)-, -CO-, -COO-, -OCO-, -OCO-O-, -S-CO-, -CO-S- or -C=C- in such a manner that oxygen atoms are not linked directly to one another, or denoting P-(Sp-X)_n-, with Sp, X and n having one of the meanings given above and P being a polymerizable group.

3. Chiral compounds according to claim 2, wherein MG is selected of formula II



II

wherein

Z is -O-, -S-, -CO-, -COO-, -OCO-, -CO-NH-, -NH-CO-, -CH₂CH₂-, -OCH₂-, -CH₂O-, SCH₂-, -CH₂S-, CH=CH-, -CH=CH-COO-, -OCO-CH=CH-, -C=C- or a single bond,

A¹ and A² are in each case independently selected from

- a) 1,4-phenylene in which, in addition, one or more CH groups may be replaced by N,
- b) 1,4-cyclohexylene in which, in addition, one or two non-adjacent CH₂ groups may be replaced by O and/or S,
- c) 1,3-dioxolane-4,5-diyl,
- d) 1,4-cyclohexenylene, 1,4-bicyclo-(2,2,2)-octylene, piperidine-1,4-diyl, naphthalene-2,6-diyl, decahydronaphthalene-2,6-diyl, or 1,2,3,4-tetrahydronaphthalene-2,6-diyl,

it being possible for all these groups to be unsubstituted, mono- or polysubstituted with halogen, cyano or nitro groups or alkyl, alkoxy, alkylcarbonyl or alkoxy carbonyl groups with 1 to 7 C atoms, wherein one or more H atoms may be substituted by F or Cl, and

m is 0, 1, 2 or 3.

4. Chiral compounds according to at least one of claims 1 to 3, wherein at least one of R¹, R², R³ and R⁴ is denoting R as defined in formula I*.

5. Chiral compounds according to at least one of claims 1 to 4, wherein R is alkyl or alkoxy with 1 to 12 C atoms.

6. Chiral compounds according to at least one of claims 1 to 5, wherein at least one of the groups R is denoting P-(Sp-X)_n-, with P, Sp, X and n having each independently one of the meanings given in formula I*.

7. A liquid crystalline mixture comprising at least one chiral compound according to at least one of claims 1 to 6.

8. A liquid crystalline mixture according to claim 7, further comprising at least one polymerizable mesogenic compound having at least one polymerizable functional group.

9. A chiral linear or crosslinked liquid crystalline polymer obtainable by polymerizing a mixture according to claim 8.

10. Use of a chiral compound, mixture or polymer according to at least one of claims 1 to 9 in liquid crystal displays, like for example STN, TN, AMD-TN, temperature compensation, guest-host, phase change or surface stabilized or polymer stabilized cholesteric texture (SSCT, PSCT) displays, in active and passive optical elements like polarizers, compensators, alignment layers, colour filters or holographic elements, in adhesives, synthetic resins with anisotropic mechanical properties, cosmetics, diagnostics, liquid crystal pigments, for decorative and security applications, in nonlinear optics, optical information storage or as chiral dopants.

11. A liquid crystal display comprising a liquid crystalline mixture according to claim 7 or 8.



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number
EP 99 11 8675

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
Y	EP 0 351 746 A (HOECHST AG) 24 January 1990 (1990-01-24) * the whole document *	1-11	C07C69/734 C09K19/58 C07D317/32 //C07M7:00
Y	DUBAL H R ET AL: "THREE CLASSES OF NEW CHIRAL DOPANTS: SYNTHESIS AND PHYSICAL QUALIFICATION AS DOPANTS FOR PRACTICAL FLC-MIXTURES" JAPANESE JOURNAL OF APPLIED PHYSICS, JP, PUBLICATION OFFICE JAPANESE JOURNAL OF APPLIED PHYSICS. TOKYO, vol. 27, no. 12, PART 02, - December 1988 (1988-12) page L2241-2244 XP000080785 ISSN: 0021-4922 * the whole document *	1-11	
D, A	EP 0 441 213 A (F. HOFFMANN-LA ROCHE AG) 14 August 1991 (1991-08-14) * the whole document *	1-11	
			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
			C07C C09K C07D
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 28 December 1999	Examiner Beslier, L
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons</p> <p>& : member of the same patent family, corresponding document</p>			

EPO FORM 1503 03.82 (P04C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 99 11 8675

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

28-12-1999

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 351746 A	24-01-1990	DE 3824902 A	15-02-1990
		AT 99304 T	15-01-1994
		DE 58906540 D	10-02-1994
		JP 1969463 C	18-09-1995
		JP 2088573 A	28-03-1990
		JP 6099422 B	07-12-1994
		KR 126469 B	24-12-1997
		NO 892983 A,B,	23-01-1990
		US 5651918 A	29-07-1997
EP 441213 A	14-08-1991	DE 59106444 D	19-10-1995
		HK 19997 A	20-02-1997
		JP 2960975 B	12-10-1999
		JP 4244077 A	01-09-1992
		US 5637255 A	10-06-1997